

NATIONAL OPEN UNIVERSITY OF NIGERIA DEPARTMENT OF PURE AND APPLIED SCIENCES SEPTEMBER, 2020_1 EXAMINATIONS

COURSE CODE: CHM 417
COURSE TITLE: INDUSTRIAL CHEMICAL PROCESSES 11
TIME: 2 HRS

INSTRUCTION: Answer question 1 and any other 3 questions

QUESTION ONE

(a) Explain Smelting method using examples and equations (6marks)

(b) Briefly discuss the major steps involved in metallurgy. (6marks)

(c) State with chemical formular the main constituent of hydraulic cement. (4marks)

(d) With the aid of chemical equation only, state how ammonium phosphate is prepared.

(4marks)

(e) Explain how HCl is produced from the chlorination process, indicating its concentration.

(5marks)

QUESTION TWO

(a) With respect to metallurgy, write short notes on

i. Reduction with carbon (3 ½ marks)

ii. Reduction of complex salts 5mks

iii. Auto reduction method (2 ½ marks)

(b) What is the importance of a regenerative system? (1mark)

(c) List the physical methods of purifying metals. (3marks)

QUESTION THREE

- (a) Mention the qualities of a good electrode. (2 ½ marks)
- (b) Discuss the industrial application of magnesium by electrolysis of Magnesia. (6marks)
- (c) Considering what happens at the cathode in the Chlor-alkali process, differentiate between the mercury cell, diaphragm and membrane cell processes. (3marks)
- (d) Using chemical equations only, show the reactions in the chlor-alkali process.

(3 ½ marks)

QUESTION FOUR

Using a schematic diagram only, describe the process used in the manufacture of Phosphate fertilizers. (15marks)

QUESTION FIVE

- (a) Discuss in detail the production and use of Urea as a fertilizer. (10marks)
- (b) Show with a well labeled diagram the Chlorination process of HCl production. (5marks)